8.33 8.31 8.10 7.86 7.76

Int = Average of normalized integrals values

MW = Molecular weight

P = Purity (as percent value)

m = mass

10.5

10.0

9.5

9.0

n = number of protons giving rise to a given NMR signal (The total number of protons is set to one because an average of all normalized integrals is carried out)

$$n_{IS} = 1$$
  $n_1 = 1$ 

$$Int_{IS} = 1.03$$
  $Int_1 = 0.99$ 

 $MW_{1S} = 165.15 \text{ g/mol}$   $MW_1 = 379.15 \text{ g/mol}$ 

 $m_{IS} = 2.53 \text{ mg}$   $m_1 = 5.69 \text{ mg}$ 

 $P_{IS} = 98\%$ 

$$P [\%] = \frac{n_{IS} \cdot Int_2 \cdot MW_2 \cdot m_{IS}}{n_2 \cdot Int_{IS} \cdot MW_{IS} \cdot m_2} \cdot P_{IS} = 95 \%$$

2.04 -

8.0

7.5

7.0

6.5

6.0

8.5

5.5

5.0

f1 (ppm)

